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Tips to Balance

Chemical Equations

Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value...

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Tip # 2: You should remember that polyatomic ions should be balanced as a whole. For instance, SO_4 should be balanced... Tip # 3: You should ...

49 Balancing Chemical Equations Worksheets [with Answers]

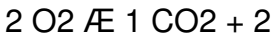
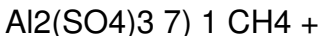
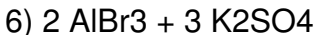
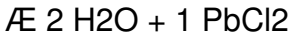
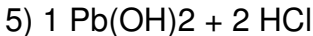
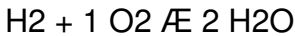
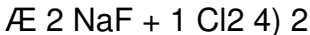
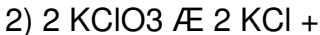
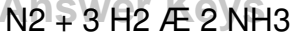
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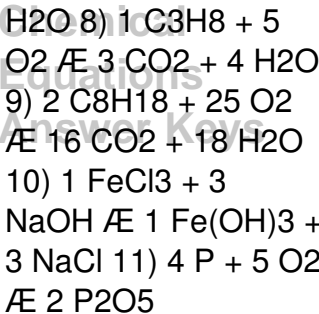
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Key Balance the

equations below: 1) 1



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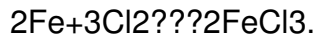
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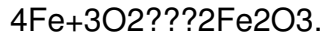
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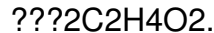
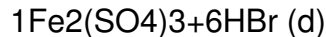
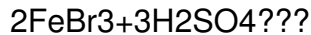
(Coefficients equal to one (1) do not need to be shown in your answers). (a)



(b)



(c)



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a)

hydrogen+iodine--->

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hydrogen iodide. b)
lithium+hydrochloric
acid---> lithium
chloride +hydrogen. h
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Interpreting chemical
formulas. Introduction:

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To balance a chemical equation, you first need to be able to count how many atoms of each element are on each side of the equation.

Student Exploration- Balancing Chemical Equations (ANSWER KEY)

Answer Key For The
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**Chemical Equations
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Anne Marie

Helmenstine, Ph.D.

Updated January 04,

2019. A balanced

chemical equation

gives the number and

type of atoms

participating in a

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reaction, the reactants, products, and direction of the reaction. Balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow.

How to Balance

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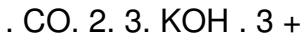
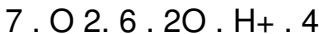
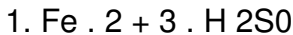
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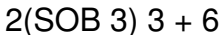
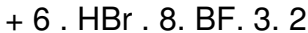
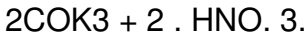
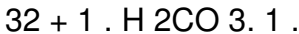
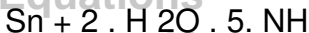
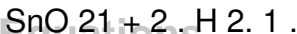
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About Chemistry <http://chemistry.about.com>

Balance the following chemical equations.



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8. Start level 2 of the balancing equation game. Take turns in your group to balance the equations in the sim, using your strategies from Level 1, and adding new strategies as needed. Each person should be in charge of balancing at least one equation, asking for

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help from the group as needed. As a group, write down the equations as you solve them. 9.

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