

## Chapter 9 Stoichiometry Review Answers

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Chapter 9 Review Stoichiometry Answer Key

Chapter 9 - Stoichiometry 9-1 Introduction to Stoichiometry Composition Stoichiometry - deals with mass relationships of elements in compounds Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction I. Reaction Stoichiometry Problems A. Four problem Types, One Common Solution

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Worksheet for Basic Stoichiometry CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H<sub>2</sub> mc06se cFMsr i-vi Q.

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### Chapter 9 Review Stoichiometry Answers

Chapter 9: Standard Review Worksheet 1. Answers will vary. An example is included below:  $2H_2O_2(aq) \rightarrow 2H_2O(l) + O_2(g)$  This describes the decomposition reaction of hydrogen peroxide. Microscopic: Two molecules of hydrogen peroxide (in aqueous solution) decompose to produce two molecules of liquid water and one molecule of oxygen gas.

### Chapter 9: Standard Review Worksheet

Stoichiometry b. Theoretically, how many moles of  $NH_3$  will be produced? PROBLEMS Write the answer on the line to the left, Show all your work in the

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space provided. 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H<sub>2</sub> according to the ...

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