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oct - oxide. Use your answers to
Questions 9-11 to write a guideline for

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how and when to modify a prefix name for a molecular compound. Come to a consensus within your group. If the second element in the name of the molecule is oxygen. drop the "a" or "o" at the end of the prefix in front of oxygen.

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18 Naming Molecular Compounds-S -
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Naming Ionic Compounds □ Answer
Key Give the name of the following
ionic compounds: Name 1) Na_2CO_3
sodium carbonate 2) NaOH sodium
hydroxide 3) MgBr_2 magnesium
bromide 4) KCl potassium chloride 5)

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FeCl₂ iron (II) chloride 6) FeCl₃ iron
(III) chloride 7) Zn(OH)₂ zinc
hydroxide 8) Be₂SO₄ beryllium
sulfate 9) CrF

Naming Ionic Compounds □ Answer
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Answers □ Naming Chemical

Compounds . Name the following

chemical compounds: 1) NaBr sodium

bromide. 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ calcium

acetate. 3) P_2O_5 diphosphorus

pentoxide. 4) $\text{Ti}(\text{SO}_4)_2$ titanium(IV)

sulfate. 5) FePO_4 iron(III) phosphate.

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6) K_3N potassium nitride. 7) SO_2 sulfur dioxide. 8) $CuOH$ copper(I) hydroxide. 9) $Zn(NO_2)_2$ zinc nitrite. 10) V_2S_5

Answers □ Naming Chemical
Compounds

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Molecular compounds unit 6

worksheet 4. Type keywords and hit
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Whippany Park High School #230344
... Naming Ionic Compounds Answer
Key - PDF #230349. Intensified
Chemistry - Units E-I & Midterm -
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Molecular compounds unit 6
worksheet 4

Steps to Naming Covalent
Compounds. First, identify the
elements present. Second, look at the
subscript of each element to determine
which prefix to use. (If an element
does not have a prefix, assume that

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the subscript is 1. Third, apply the above naming scheme.

Steps to Naming Ionic and Covalent
Compounds - Owlcation ...
Naming Molecular Compounds
Worksheet Answers Pogil Activities
For High School Chemistry We found

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Answer Key about Naming Molecular
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Naming Molecular Compounds 5
Extension Questions 19. This activity

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focused on molecular (covalent) compounds, while an earlier activity.

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To demonstrate steps 1 and 2a, HCl is named hydrogen chloride (because hydrogen is to the left of chlorine in the periodic table), and PCl₅ is phosphorus pentachloride. The order of the elements in the name of BrF₃,

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bromine trifluoride, is determined by the fact that bromine lies below fluorine in Group 17.

3.6: Naming Covalent Compounds -
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Key Points In nomenclature of simple molecular compounds, the more

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Answer Key If 8766
electropositive atom is written first and the more electronegative element is written last with an -ide suffix. The Greek prefixes are used to dictate the number of a given element present in a molecular compound.

Naming Compounds | Boundless

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Chemistry Key If8766

If a binary compound is composed of two nonmetals, it is a covalent/molecular compound. Use the appropriate prefixes to name these simple compounds. Mono- is only used as a prefix on the second element in the formula. The ending for

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the second element is always removed and replaced with -ide. 1 = mono- 2 = di- 3 = tri- 4 = tetra- 5 = penta- 6 = hexa- 7 = hepta- 8 = octa- 9 = nona- 10 = deca-.

Naming Binary Covalent Compounds
Flashcards | Quizlet

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What is a similarity in naming ionic and molecular compounds? (the second element ends in -ide) Ending class this way allows me to get at the key learning objectives of the class. Then, depending on how much students completed, I show some of the answers from the Naming molecular

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and ionic compounds key and the
Modeling compounds answer key. so

...

Eleventh grade Lesson Ionic vs.
Molecular Compounds
Ionic and Covalent Compounds Name:
KEY!! 1. We differentiate between two

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types of compounds: IONIC and COVALENT. ! 2. Ammonia, NH_3 is a COMPOUND while nitrogen and hydrogen are ELEMENTS. ! 3. In general, molecular compounds form when NONMETALS combine together. ! 4. In general, ionic compounds form when METALS &

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NONMETALS combine together. ! 5.

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