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The book concentrates on core topics which are most likely to be common to those organic chemistry courses which follow on from a foundation or introductory general chemistry course. Description : This book attempts to condense the essentials of organic chemistry into a manageable text of 310 pages which is student friendly and which does not cost an arm and a leg.

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Organic Chemistry Exam 1 study guide Chapter 1 Chemical

bonding and chemical structure 1.2 Lewis structures Draw

molecule, show all valence electrons, add electrons to give non-

hydrogen atoms full octets Formal charge = (Valence electrons) -

(nonbonding electrons) - (½ bonding electrons) Nonpolar when

electrons are shared evenly More eneg in top right (More eneg,

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more attraction towards other atoms) Molecular dipole - Sum of individual dipoles 1.3 Bond length decreases left to ...

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