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Thermodynamic identity (Schroeder 3.4) $dU = TdS - pdV$ (1.5) Total differential

When we have a total differential, the things in front of the dS, dV , etc are partial derivatives. $T = \left(\frac{\partial U}{\partial S}\right)_V - p = \left(\frac{\partial U}{\partial V}\right)_S$ (1.6) Also, you can integrate along a path using a total differential, and can do linear algebra using total differential equations, e.g. substituting one for another.

Thermal Physics - Oregon State University

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Schroeder An Introduction To Thermal Physics

Zero-point energy (ZPE) is the lowest possible energy that a quantum mechanical system may have. Unlike in classical mechanics, quantum systems constantly fluctuate in their lowest energy state as described by the Heisenberg uncertainty principle. As well as atoms and molecules, the empty space of the vacuum has these properties. According to quantum field theory, the universe can be thought ...

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